

## Chemistry Chapter 9 Stoichiometry Test

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### Chapter 9 Stoichiometry Multiple Choice Answers

Modern Chemistry Chapter 9 Stoichiometry Test Answers File Type answer on the line to the left Show all your work in the space provided 1 \_\_\_\_\_ The actual yield of a reaction is 22 g and the theoretical yield is 25 g Calculate the percentage yield 2 60 mol of N<sub>2</sub> are mixed with 120 mol of H

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Practice Test Ch3 Stoichiometry (page 3 of 3) 1. d It might be easiest to balance the equation with mostly whole numbers: 2 NH<sub>3</sub> + 7/2 O<sub>2</sub> → 2NO<sub>2</sub> + 3H<sub>2</sub>O. The question asks about the amount of oxygen reacting with ONE mole of ammonia, thus cut the 7/2 (3.5) of oxygen in half to 1.75 2. c Balance: CH

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To understand stoichiometry, start with this introduction to the topic. It might also help to review molecules and moles, which includes how chemical formulas work. Ready for another quiz? Here's a quick self-test about the mole. If you'd rather switch gears, see if you know the answers about how chemistry explains the real world.

### Practice Test Ch 3 Stoichiometry Name Per

CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation: C<sub>3</sub>H<sub>4</sub>(g) + xO<sub>2</sub>(g) → 3CO<sub>2</sub>(g) + 2H<sub>2</sub>O(g) 4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of C<sub>3</sub>H<sub>4</sub>? 2 mol O<sub>2</sub>:1 mol H<sub>2</sub>O c. What is the mole ratio ...

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equation. Chapter 9 Stoichiometry Table of Contents Chapter 9 Stoichiometry Section 1 Chapter 9 Section 1 Intro to Stoichiometry. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Blair12\_Armstrong. Key Concepts: Terms in this set (10) Stoichiometry is the branch of chemistry that deals with elements in compounds ...

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