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Chapter 2 Properties of Matter Section

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2.3 Chemical Properties (pages 54–58)

This section discusses chemical properties and describes clues that may show that a chemical change has taken place. Reading Strategy (page 54) ...
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physical 116 Guided Reading and Study
Workbook. Chemical reactions occur
when atoms are separated, joined, or

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rearranged, however, atoms of Chapter
4

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114 Guided Reading and Study
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Reactions(continued) 9. Use the symbols

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in Table 11.1 on page 323 to write a skeleton equation for the following chemical reaction. Hydrochloric acid reacts with zinc to produce aqueous zinc(II) chloride and hydrogen gas. Balancing Chemical Equations (pages 324–328) 10.

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CHAPTER 9, Chemical Names and
Formulas(continued) 8. Use the periodic
table to write the name and formula
(including charge) for each ion in the
table below. Polyatomic Ions (pages
257-258) 9. What is a polyatomic ion?
10. Is the following sentence true or

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false? The names of polyatomic anions
always end in ...

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Answers

Waves.47 10. What is the highest atomic number shown on the periodic table? 10
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chemical bonds form, valence electrons
are either or between atoms. 17. A way
to show the number of valence electrons

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an atom has, using dots ...

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CHAPTER 10, Chemical

Quantities(continued) 9. Complete the table about representative particles and moles. The Mass of a Mole of an Element

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Answers

(pages 293-294) 10. What is the atomic mass of an element? 11. Circle the letter of the phrase that completes this sentence correctly. The atomic masses of all elements

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letter of each sentence that is true about
Dalton's atomic theory. Chapter 4

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about genes.

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Chapter 8 Photosynthesis

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The chemical formula of a binary ionic compound includes the cation's symbol, followed by the anion's symbol, with subscripts that balance positive and negative charges. Compounds With Polyatomic Ions Ionic compounds are named by joining the cation

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element chemical bond The study of the
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and types of atoms represented by these chemical formulas. a. Fe_2O_3 b. KMnO_4 c. CH_3 d. NH_4NO_3 7. What is a formula unit? 8. Explain why the ratio of magnesium ions to chloride ions in MgCl_2 is 1:2. 9. Describe the ...

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Glucose Glycolysis Cytoplasm ATP

Electrons carried in NADH Mitochondrion

Electrons carried in NADH and FADH₂

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Pyruvic acid Krebs Cycle Electron

Transport Chain ATP ATP Name _____

Class _____ Date _____

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Chapter 10

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to Earth Science ... Earth Science Guided Reading and Study Workbook ... Chapter 2 Minerals Summary 2.1 Matter An element is a substance that cannot be broken down into simpler substances by chemical or physical means. An atom is the smallest particle of matter that ...

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