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Chapter: The Periodic
Law Use the periodic
table below to answer
the questions in this

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Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

_____ 1.

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Distinguish between ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided.

The molar enthalpy of fusion for water is 6.008 kJ/mol. 28.

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Percentage Yield.
Terms in this set (11)
Reaction stoichiometry.
calculations involving
the mass relationships
between reactants and
products in a ...

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Chapter: The Periodic
Law PART I In the

space provided, write
the letter of the term
or phrase that best
completes each
statement or best
answers each question.

_____ 1. In his periodic
table, Mendeleev did
not list all of the
elements in order

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Chapter 12 (pages 320
- 321) - Answer
questions - 1-9, 14.

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Ch. 1 - #5, 6, 24 5.

volume is the amount
of space taken up by
an object, and mass is
the amount of matter
an object contains. 6.

Area is the amount of
surface an object has,
and density is the
amount of matter in a
given volume. 24. a.

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_____ 21. According to Einstein, which of the following can behave like a wave and also like a stream of particles? a. a noble gas b. the atomic nucleus c. electromagnetic radiation d. a hydrogen atom in the ground state _____ 22.

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PLAY. The basis of an

ionic bond is the__

absorption of water

into their solid

structures. Elements

tend to gain or lose

electrons in order to

acquire the electron

configuration of a__

noble gas. Ions that are

formed from more than

one atom are called__

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Stoichiometry PART I

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. 1.

Knowing the mole ratio of a reactant and product in a chemical reaction would allow you to determine a. the energy released in the reaction. b ...

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