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CHAPTER 10 REVIEW States of Matter
SECTION 3 SHORT ANSWER Answer the
following questions in the space
provided. 1. Match description on the
right to the correct crystal type on the
left. b ionic crystal (a) has mobile
electrons in the crystal c covalent

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molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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jelizondo7. The parentheses are just

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there so you can choose ignore the stuff in parentheses and make the space race much easier. So just ignore them in general. Terms in this set (34)

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22O 11 10 atoms c. Fe(ClO₂)₃ 9 atoms
d. Fe(ClO₃)₂ 5. N₂O₅ a. What is the
formula for the compound dinitrogen
pentoxide? iron(II) oxide b. What is the
Stock system name for the compound

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FeO? H₂SO₃ c. What is the formula for sulfurous acid? phosphoric acid d. What is the name for the acid H₃PO₄?

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Chemistry CHAPTER 22 REVIEW Organic
Chemistry SECTION 1 SHORT ANSWER
Answer the following questions in the
space provided. 1. Name two types of
carbon-containing molecules that are
not organic. _____ 2. _____ Carbon atoms
form bonds readily with atoms of

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Answers

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~~Chapter 22~~

CHAPTER 5 REVIEW The Periodic Law
SECTION 1 SHORT ANSWER Answer the
following questions in the space
provided. 1. c In the modern periodic
table, elements are ordered (a)
according to decreasing atomic mass.
(b) according to Mendeleev's original

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design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

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~~7-Chemical Formulas and Chemical
Compounds~~

Chapter 22-Assignment E: Summary and Review In Chapter 6, we urged you to learn the system when studying inorganic nomenclature. Memorize the minimum needed. In organic chemistry, that minimum is the list of functional

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groups. A functional group summary is given in Table 22.5, Section 22.14. To what are functional groups attached?

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Modern Chemistry 90 Chapter Test
Name Class Date Chapter Test B,
continued PART III Match the letter on

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the diagram with the term for that point or region. 18. Critical point 19. Vapor 20. Normal freezing point 21. Triple point 22. Solid 23. Normal boiling point 24. Liquid 25. Critical pressure PART IV Write the answers to the following ...

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HOMEWORK 22-2 (pp. 702-704)

VOCABULARY Write true or false. 1. The band of stability represents the stable nuclei cluster over a range of electron-proton ratios. _____ 2. The stability of a nucleus is greatest when the nucleons are in a 1:1 ratio. _____ 3. 2, 20, 50, and 126 are all magic numbers. _____ 4.

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~~5 The Periodic Law~~

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