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## **The Basics of General, Organic, and Biological Chemistry**

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In chemistry, a solution is a special type of homogeneous mixture composed of two or more substances. In such a mixture, a solute is a substance dissolved in another substance, known as a solvent. The mixing process of a solution happens at a scale where the effects of chemical polarity are involved, resulting in

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interactions that are specific to solvation.

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Introduction. A simple demonstration of how a precipitate is evidence of a chemical reaction taking place is performed by mixing solutions of calcium chloride and sodium carbonate to form the precipitate calcium carbonate ( $\text{CaCO}_3$ ).  $\text{CaCl}_2(\text{aq}) + \text{Na}_2\text{CO}_3(\text{aq}) \rightarrow \text{CaCO}_3(\text{s}) + 2\text{NaCl}(\text{aq})$ . To Conduct Demonstration

## **Na<sub>2</sub>CO<sub>3</sub> - Lecture Demonstration Manual General Chemistry**

The purpose of this lab is to synthesize cyclohexanone. Cyclohexanone is used as a precursor for nylon. This makes it one of the largest mass produced chemicals in the industry. Billions of kilograms of cyclohexanone are produced each year for the making of nylon [1]. The synthesis of cyclohexanone ...

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